## (Closed book, no models)

Name (Please print) Last First Middle initial
Last 4 digits of Students ID Number: $\qquad$
TA $\qquad$

1) This exam consists of 5 pages (including this page) and 6 questions.
2) Do not turn the pages until 3:10 p.m.
3) Turn in your exam to one of the teaching staff by $4: 00$ p.m.
4) Put your 4 digits ID in the upper right corner of the next 4 pages. Be sure you have pages 2-5.
5) Please answer clearly in the spaces provided. The back of the page is for scratch work only.
6) Any request for re-grading should be done according to the TA's instruction.


| Page | Score |
| :--- | :--- |
| 2 | $/ 22$ |
| 3 | $/ 32$ |
| 4 | $/ 36$ |
| 5 | $/ 10$ |
| Total | $/ 100$ |

$\qquad$

1. (8pts) Name the following in IUPAC:
(a)

(Z)-1-chloro-1-iodo-2-methyl-1-butene
(b)


2-methyloct-1en-7-yne
2. (a) 4 pts. Rank the following compounds in order of increasing acidicity.

A

B

C

D

$$
\mathrm{B}>\mathrm{D}>\mathrm{C}>\mathrm{A}
$$

most acidic least acidic
(b) 2 pts. Circle the least polar compound.



(c) 5 pts. Rank the following compounds in order of increasing stability.

A

B

C

D

E
$\mathrm{E}>\mathrm{X}>\mathrm{B}>\mathrm{D}>\mathrm{A}$
most stable
(d) 3 pts. Arrange the following alcohols in order of their reactivity toward acid-catalyzed dehydration (with the most reactive first).


A


B

most reactive
$\mathrm{B}>\mathrm{X}>\mathrm{A}$
least reactive
$\qquad$
3. (32pts) Please give the expected major product(s) for the following reactions. Unless mentioned otherwise, you can assume that all reagents are present in one mole. If there is no reaction, please write NR. Show stereochemistry where necessary.
(a)



(b)


(c)


 -2 for missing
stereochemistry
(d)




(c)



 $\xrightarrow[\mathrm{CCl}_{4}]{\mathrm{Br}_{2}}$

(e)


(f)

1. $\mathrm{Hg}(\mathrm{OAc})_{2}, \mathrm{CH}_{3} \mathrm{OH}$,
2. $\mathrm{NaBH}_{4}, \mathrm{CH}_{3} \mathrm{OH}$
(g)

3. $\mathrm{BH}_{3}, \mathrm{THF}$
$\longrightarrow$
4. $\mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{NaOH}$, $\mathrm{H}_{2} \mathrm{O}$


(h)

$\xrightarrow[\text { 2. } \mathrm{NaHSO}_{3} / \mathrm{H}_{2} \mathrm{O}]{\text { 1. } \mathrm{OsO}_{4}, \text { pyridine }}$

no stereochemistry is fine, but has to be syn addition
$\qquad$
5. (16pts) Write a complete, step-wise, detailed mechanism for the following reaction.







molozonide


carbonyl oxide

6. Please show how you would carry out the following transformations.
(a)



(12pts)
$\downarrow$ MCPBA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$


